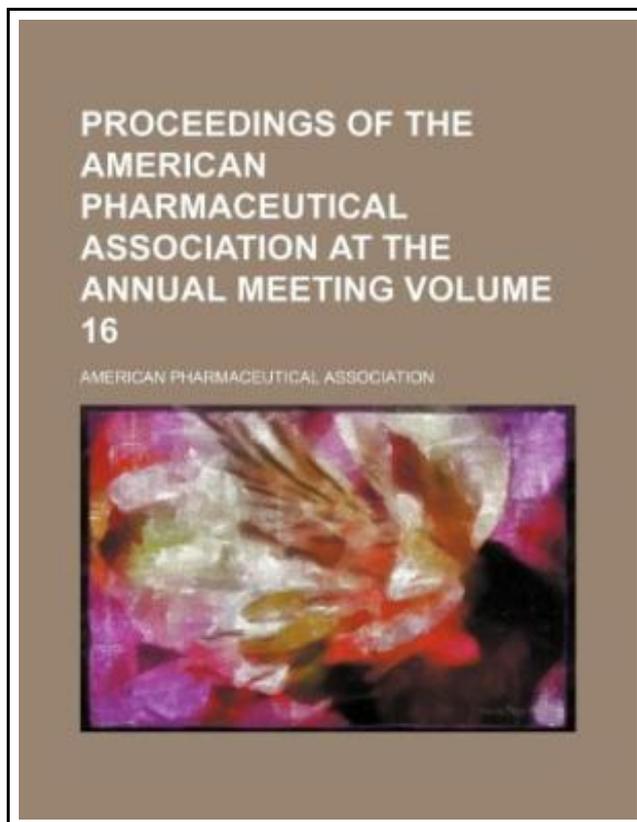


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1869 Excerpt: .the odorous compounds happen to contain sulphur, is prepared by boiling litharge in a tolerably concentrated solution of caustic soda, by which a clear solution is formed containing considerable oxide of lead, apparently playing the part of an acid. Ch. N. Jan. 10, 1868. A. J. Ph. xl, 277. THALLIUM. Thallium is extracted by WShler from flue dust by the following process: The material is repeatedly extracted by boiling water, slightly acidulated with SO₃, and filtered; the filtrate is precipitated with HCl; the chloride formed is converted into sulphate, and the latter in aqueous solution reduced by zinc or an electric current derived from a single cell. The metal is finally fused under a layer of cyanide of potassium. Ch. N. Aug. 9, 1867. Carstanjen has obtained alloys of this metal with As, Sb, Zn, Cd, Bi, Pb, Sn, Cu, Hg, Al, Mg, Na, K, all of which are described in J. p. Ch. ciii, 65-89. Thallium Amalgam. According to Regnault, thallium unites with mercury with evolution of heat, forming an amalgam which is electro-negative, compared with thallium; another proof of the rule that, when a metal dissolves in Hg with evolution of heat, the metal is electro-positive in comparison with its amalgam. Compt. rend. lxiv, 611. Ch. N. Aug. 9, 1867. Double Chloride of Thallium and Iron is formed, according to Wohler, when recently precipitated moist chloride of thallium is added to a concentrated solution of sesquichloride of iron containing an excess of strong HCl....

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